# Packing Fraction Measurement

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# **Goal and Challenges**

- Goals:
  - Measure mass and volume of Ammonia in target cups
  - $\circ$  Use packing fraction in LN<sub>2</sub> to extrapolate packing fraction in LHe
- Challenges:
  - Little data on density of Ammonia at low temperatures
  - Evaporation of LN<sub>2</sub> during measurement
  - Precise measurement of displacement
  - Precise measurement of mass



### Steps necessary

- 1. Calibrate position measurements to volume measurements in measurement container (do with water)
- 2. Establish evaporation rate of LN2 in measurement container (used for mass and volume measurements)
- 3. Using plastic beads, measure density using both water and LN2 to confirm method
- 4. Measure mass and volume of target cup in LN2

#### Procedure

- 1. Put target cup in holding container to thermalize and reduce boiling during measurement
- 2. Measure level of LN2 in measurement container, mass of measurement container and LN2, and start clock
- 3. Place target in measurement container and measure level of LN2 and mass of container
- 4. Using elapsed time and evaporation rate, use level and mass measurements to find volume and mass of target cup.

# To Do:

- Make a prototype as proof of concept
- New scale?
- Undergrad for the summer?